The cancer predisposition revolution

How was the inherited basis of cancer foreshadowed?

By David Malkin, Judy E. Garber, Louise C. Strong, Stephen H. Friend

studies of rare cancer predisposition syndromes often lead to the identification of genes critical to carcinogenesis. In 1969, Li and Fraumeni described a constellation of various cancers in the family members of four unrelated children who were diagnosed with soft tissue sarcomas. They posited that the cancers best fit an autosomal dominant pattern of inheritance, attributable to a genetic defect. At that time, cancer was not generally thought of as a genetic disease. Their hypothesis set the stage for establishing germline mutations in the tumor suppressor gene TP53 as the underlying genetic event in Li-Fraumeni syndrome (LFS) families. It also foreshadowed dozens of discoveries, still ongoing, that associate mutations in tumor suppressor genes, activated oncogenes, mitochondrial genes, and DNA repair genes with cancer predisposition phenotypes in which multiple different neoplasms occur across generations.

What makes the prescience of Li and Fraumeni remarkable is how little was known at the time. Their observation preceded both Knudson’s “two-hit” theory of carcinogenesis and the technical ability to look for heritable mutations in genes, and it was not until 1986 that the first cancer susceptibility gene, RB1, was shown to be responsible for retinoblastoma, a rare heritable cancer. In 1979, two groups discovered the p53 oncprotein, a rare heritable cancer. The field then exploded with seminal papers that paved the way for the ultimate discovery of the link between p53 and LFS: the “classic” clinical features of LFS were defined; inactivating somatic TP53 mutations were discovered in a wide spectrum of cancers; and a Trp53 transgenic mouse was created that facilitated further research.

Remarkably, well into the 21st century, not only do new genes continue to be discovered to account for long-known cancer syndromes (e.g., protection of telomeres 1 [POT1]; partner and localizer of BRCA2 [PALB2]), but new syndromes also continue to be defined. These include biallelic mismatch repair deficiency, in which early-onset cancers occur as a result of a perfect storm of inherited biallelic microsatellite gene mutations followed by somatic inactivation of a DNA polymerase (IO), and DICER1 syndrome, in which an array of childhood and adult-onset tumors are caused by inactivation of a gene that is essential for microRNA processing (12). Thus, systematic clinical cancer epidemiology, as established by Li, Fraumeni, and Miller in the 1960s, continues to influence the discovery of cancer syndromes and cancer susceptibility genes and inform our understanding of the fundamental biology of human cancer.

Although different TP53 mutations confer different degrees of penetrance, the overall lifetime risk of cancer is at least 75% in males and approaches 100% in females; the risk of developing multiple cancers is higher than in the general population by a factor of 85 (12). Despite progress in understanding the central role played by wild-type p53 in maintaining genome stability, and that of mutant p53 in cellular transformation, it is still nearly impossible to prevent or delay cancer in LFS, to predict age of onset, likelihood, or type(s) of cancers that will develop, to reduce the incidence of subsequent malignancies, or to optimally treat the cancers once they occur. Radiation-free clinical surveillance protocols may empower families to detect cancers early, offering hope for improved survival (13). However, the availability of whole-body magnetic resonance imaging (MRI) is limited in some jurisdictions because definitive evidence of efficacy is not yet available—a challenging problem in a rare syndrome. Whole-body MRI is being studied in many other cancer syndromes as well, particularly when cancer risk in childhood is a feature. Intensive screening efforts without risk reduction are fraught with additional challenges, including the establishment of detection precision, demonstration of definitive improvement in outcomes such as prolonged survival, and the complex psychological risk-benefit considerations of frequent examina-
POLYMER SYNTHESIS

Organic photocatalysts for cleaner polymer synthesis

Metal-free catalysts enable synthesis of polymers for biomedical and electronics applications

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The material properties of synthetic polymers can be tuned by changing their chain length and branching and the way in which monomer units repeat. For example, high-density polyethylene, which has little chain branching, is a stiff polymer used for food containers and drain pipes, whereas low-density polyethylene, which has more chain branching, is flexible and used to make grocery bags and bottles for chemicals.

Polymers are usually made through thermal polymerization, but recent efforts focusing on green chemistry have led to a push toward using solar energy to drive chemical reactions. Page 1082 of this issue, Theriot et al. (1) report on metal-free visible-light photocatalysts that produce well-defined polymers free of metal contamination through radical polymerization.

Free-radical polymerization, in which a thermally decomposing radical initiates the addition of one monomer unit to another, is commonly used in industry. However, this approach cannot control chain length because of the rapid termination of growing chains. To ensure perpetual growth of polymer chains with homogeneous chain lengths, atom-transfer radical polymerization (ATRP) is commonly used (2–4). Unlike free-radical polymerization, ATRP generates chains with excellent chain-end functionality, which enables reactivation for further monomer addition or even postpolymerization modification. Moreover, atom-transfer radical polymerization provides the means to generate polymers with predetermined molecular weight, narrow molecular weight distribution, and copolymer composition (5). By controlling these properties, a new range of high-value applications have emerged in diagnosis, nanomedicine, and nanotechnology (6).

Nevertheless, ATRP has required transition-metal catalysts, predominantly copper halides, which become part of the polymer. The Cu(0) state activates polymerization, whereas Cu(II) deactivates it, but unavoidable chain termination can lead to accumulation of Cu(II). Thus, relatively high Cu concentrations—10,000 parts per million (ppm)—are required to maintain the equilibrium between Cu(0) and Cu(II). Alternative solutions, specifically ARGET (activators regenerated by electron transfer) atom-transfer radical polymerization, have been proposed to reduce the amount of Cu (from 10,000 to 10 ppm) through the use of organic reducing agents, such as ascorbic acid and glucose, that maintain the Cu(0) and Cu(II) equilibrium (2). Alternatively, the use of ion-exchange resin and absorbent, such as alumina, silica, or talc, can further reduce the concentration of catalyst in the final polymer product (5).

Although Cu contamination can be minimized by these approaches, complete removal of this transition metal is necessary for applications involving microelectronics and biomaterials, which has spurred the development of metal-free catalyst systems. The recent work of the groups of Hawker and Matyjaszewski introduced control over atom-transfer radical polymerization through electrochemistry (7) and photochemistry [for selected examples and a review, see (8–10)]. Remarkable spatial, temporal, and sequence control has enabled fine tuning of the properties of the generated materials.

Realizing the potential of photochemical polymerization, Miyake and Theriot (II) initially explored the use of perylene as a photocatalyst for activation of photo-atom-transfer radical polymerization under visible light.
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